

Amalgamation of Nanodiamond and Epoxy

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Abstract

Nanoscale diamond particles are known as nanodiamond (ND). Nanodiamond has attracted much attention of the polymeric composite community. Owing to its superior properties, ND holds great potential to improve tribological characteristics of the composites. In epoxy/nanodiamond nanocomposite, large surface to volume ratio of ND results in dramatic increase in volume of inter-phase, i.e. polymer volume which is close enough to nanodiamond. This, in turn, allows nanodiamond to exert significant impact on epoxy properties even at low concentrations. It is critically important to have physical/covalent bond at epoxy/ND interface from nanoparticle surface to the macromolecules of the matrix. Mechanical properties, thermal stability and electrical conductivity of the covalently compatibilized epoxy/ND composite have been found better than neat epoxy matrices and compatible composites. Epoxy/ND composite has huge range of technical applications ranging from biomedical—to—electronics—to—aerospace.

Keywords: Epoxy; Nanodiamond; Interface; Nanocomposite; Application

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1. Introduction

Nanodiamond (ND) forms a significant class of nanomaterials. ND is about 5 nm in diameter. Nanodiamond not only has exceptional features of diamonds such as hardness, stiffness, chemical stability, and strength, but also has the significance of nanomaterials such as large surface area and other physical and structural properties [1]. In 1963, nanodiamond was discovered in explosive mixtures in immense concentration. The mass production of ND was started in 1990s [2]. Nanodiamond covers miscellaneous areas of research. A very significant use is found in polymer/nanodiamond composite. ND possesses marvellous physical and chemical properties relative to the conventional constituents. Different methods have been used to prepare high performance polymer/ND composite [3-5]. The main focus of these methods is to achieve efficient dispersion of detonation nanodiamond particles in amorphous thermoplastic/thermoset matrices. For estimation of polymer/nanodiamond properties, various microscopic and physical techniques have been used. In polymer/nanodiamond composite, mechanical properties, thermal features, and large surface to volume ratios are important parameters. ND reinforcement in matrices has led to ultra-high stiffness and hardness of polymer/nanodiamond nanomaterial [6-8]. The mechanical properties of polymers reinforced with nanodiamond nanofiller have been studied. Even at very small content (~1 wt.%), mechanical properties have been increased up to 100 % [9, 10]. Both physical as well as chemical bondings have been observed in these nanocomposites. However, the exact mechanism of reinforcement in the polymer/ND composite is still unknown. The synergy between the nanofiller content and polymer properties has also been stressed. Polymer/nanodiamond composite have been employed in several structural applications, biomedical, magnetic sensors, etc. In this article, a brief but comprehensive coverage of nanodiamond—to—polymer/nanodiamond materials has been conversed.

2. Nanodiamond

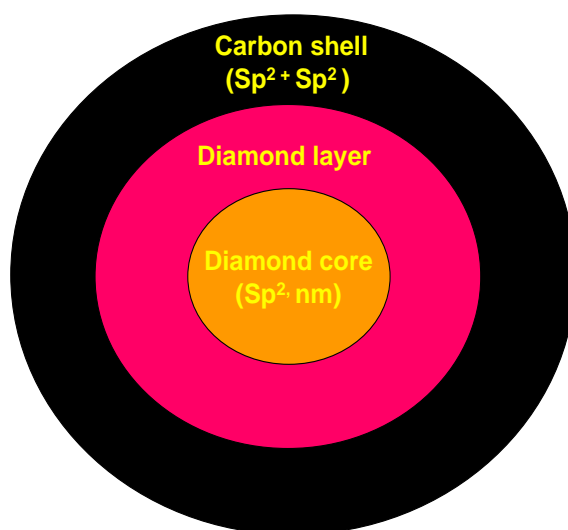


Fig. 1 Nanodiamond structure

Nanodiamond (ND) is a member of nanocarbon family comprising diamondoids, fullerenes, tubes, onions, horns, cones, rods, and several other forms [11-13]. The electronic configuration of carbon is $1s^2, 2s^2, 2p^2$. In graphite, carbon atom has sp^2 hybridization to form two dimensional planar hexagonal structure. On the other hand, tetrahedral structure of diamond is formed by sp^3 hybridization. The carbon layer stacking form modified face centered cubic structure [14]. The structure of ND particle consists of a central part of diamond core made up of sp^3 hybridized carbon, core is covered by shell of sp^2 hybridized carbon, and outer surface with functional groups formed by carbon atoms (Fig. 1) [15]. Initial studies on ND were performed in 1960s. However, ND was unknown to the world till 1980s. In late 1990s, number of significant breakthroughs led to immense attention towards nanodiamond in materials technology. The most commonly used method for the production of nanodiamond (large scale) is the detonation synthesis. Detonation of carbon containing explosive may produce small sized particles of ~ 4 -5 nm. Nanodiamond usually has narrow size distribution and facile surface functionalization capability. ND surface comprises various hydrocarbon functional groups such as carboxylic, ether, oxiran, and carbonyl groups [16]. Consequently, ND has intrinsic hydrophilic surface to be employed in biomedical applications. ND is excellent contender for surface functionalization Due to the nano-size, large and open surface area (300 - $500 \text{ m}^2/\text{g}$), and characteristics for purification with oxidizing agents [17]. A simple reaction of epoxide monomer with acid functional nanodiamond is given in Fig. 2.

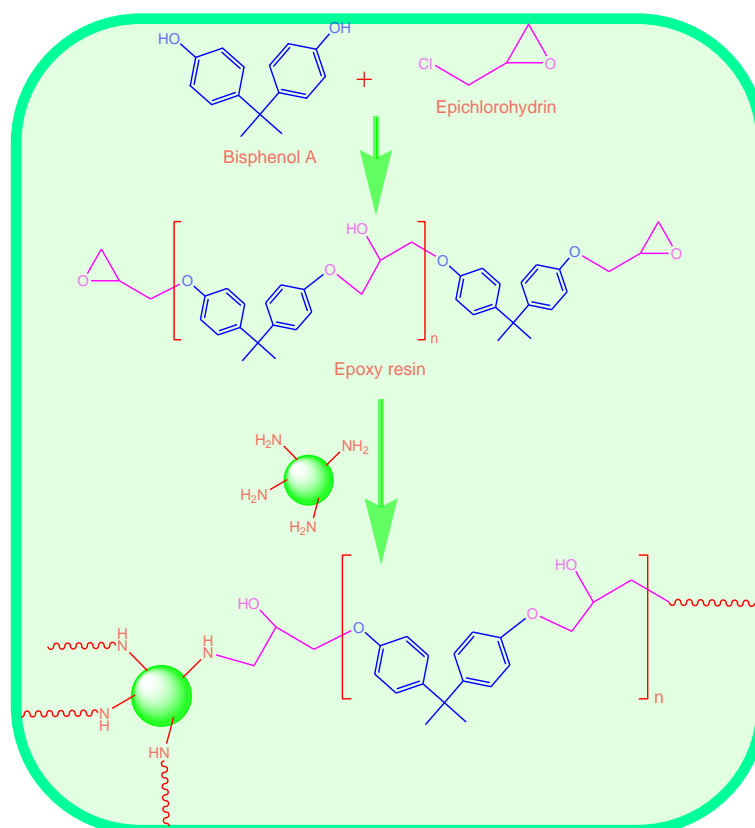


Fig. 2 Chemical reaction between epoxy and functional nanodiamond.

3. Polymer and nanodiamond composite

Nanodiamond particles endow superior hardness, Young's modulus, electrical resistivity, high thermal conductivity, chemical stability, and biocompatibility to polymer nanocomposites. Consequently,

excellent features make ND an excellent candidate for polymer reinforcement. The influence of ND on mechanical features of polymers and rubbers has been investigated [18, 19]. Polyvinyl alcohol (PVA) and nanodiamond composite have been studied [20]. The matrix was reinforced with up to 7 wt.% ND. The stiffness and strength of PVA/ND were found to increase, however ductility was decreased with large ND concentration. Polyamide and ND composites were also reported. There was considerable increase in the Young's modulus and hardness of ~20wt. % ND loaded polyamide fibers [21]. Properties of polyacrylonitrile and polyamide 11 nanofibers have also been improved with the addition of ND content. Poly(methyl methacrylate) (PMMA) and modified ND have been used in for improvement of mechanical properties in dental implants. Polycarbonate (PC) has also been reinforced with pure and functionalized ND. PolyLactic acid composites has also shown improved storage modulus, tensile modulus, and the tensile strength with 5 wt.% ND loading [22]. Similarly, several other type of polymer/ND composite has been reported [23-32].

4. Epoxy/nanodiamond amalgamation

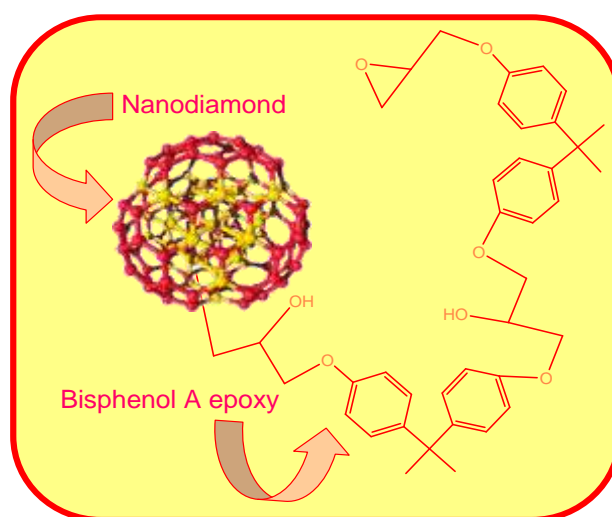


Fig. 3 Covalent interaction between epoxy chains and nanodiamond.

A common thermosetting polymer is epoxy, which is used as matrix material for composites in building materials, adhesives, composites, laminates, and coatings due to low cost, adhesion properties, and outstanding mechanical and thermal resistance. Epoxy resins are also known as polyepoxides. These are reactive prepolymers/polymers with epoxide groups. Epoxy resins may be reacted or cross-linked using various catalysts, diamines, polyfunctional amines, multi-functional acids, and alcohols. Reaction of polyepoxides with themselves or with polyfunctional hardeners forms a thermosetting polymer. Several nanofillers have been investigated to reinforce epoxy systems. Polyamide, poly(vinyl alcohol), polyisoprene, and poly(L-lactic acid) composites have been reported with nanodiamond. So, it is mostly reinforced in thermoplastic matrices [33]. Epoxy/ND composites have been prepared using various routes such as simple solution phase method, melt mixing, in-situ technique, electrospinning, and several other methods. Epoxy/ND composites with nanofiller loading demonstrated significant increase in glass transition temperature and other physical properties. On the other hand, the composites of ND with thermosetting polymers have been less studied. Mechanical behaviour of epoxy/ND composites has been studied. Storage modulus, Young's modulus, and

hardness have been investigated for epoxy/ND nanocomposite [34]. Enhanced mechanical features have been observed at low ND content. Epoxy/ND composites have been considered significant for engineering applications. The curing of epoxy has been achieved by mixing with various hardeners and cross linkers [35]. Thermal conductivity of epoxy/nanodiamond composites with nanodiamond concentration (0-25 vol.%) have also been widely studied in addition to the mechanical properties [36]. The nanodiamond may interact chemically with the epoxy chain as shown in Fig. 3.

Epoxy/nanodiamond derivatives have also been synthesized using grafting epoxy monomers onto the surface of nanodiamond. The materials have been characterized by thermogravimetric analysis (TGA) and Fourier transform infra red (FTIR) spectroscopy. The ratio of grafted epoxy groups was determined by TGA. The epoxy functionalization of nanodiamond material enables variety of advanced engineering and biomedical applications. Amine-functional ND-NH₂ has also been reacted with the epoxy resin. The formation of strong covalent epoxy/ND interface was identified by differential scanning calorimetry (DSC). Tetrahydrofuran (THF) was used as solvent. ND-NH₂-epoxy composites with uniformly dispersed particles exhibited improved Young's modulus for the compatible nanocomposite [37, 38]. Similarly, extensive tribological studies have been performed on epoxy/ND nanocomposite. The materials showed improved properties at very low nanodiamond content. Addition of only 0.1 wt.% ND may decrease the coefficient of friction and the wear rate, however increase the tensile strength, toughness and hardness of the material.

5. Significance of epoxy/nanodiamond composite

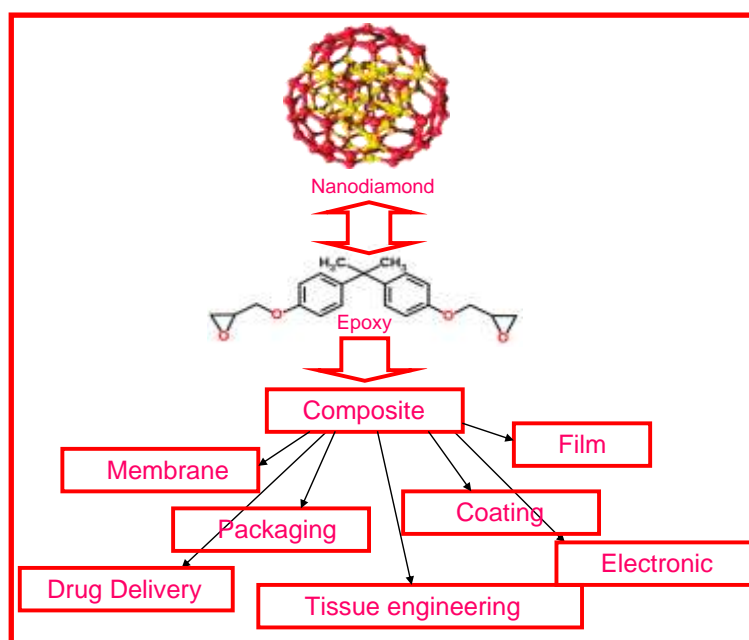


Fig. 4 Applications of epoxy/ND nanocomposite.

Applications of multifunctional epoxy/ND have been focused (Fig. 4). Epoxy/ND composites exhibit unique combination of properties for electronics, packaging, membranes, and biomedical applications such as tissue engineering scaffolds, drug delivery, and biomedical polymer devices with ND. Electromagnetic compatibility is essential in functioning of electrical devices. Moreover, it is

important to shield the electrical devices (integrated circuits, processors, and equipments) against electromagnetic interference. Consequently, epoxy/ND nanocomposites have been used as shielding materials. Epoxy/ND has excellent electromagnetic absorption features due to high electronic conductivity and high defective inner shells [39]. Thermal conductivity and mechanical features of nanodiamond strengthened epoxy matrix composites have also been studied [40]. Young's modulus and hardness of these composites were found to be improved up to several folds [41]. The bonding between epoxy and ND particles may result in enhanced thermal conductivity of these composites. However, agglomeration in epoxy/ND composites may cause damage to efficient hardness of the material [42, 43]. The properties were found better than the other polymer/carbon nanofiller composite [44-50].

6. Conclusion

Nanodiamond presents several advantages to epoxy nanocomposites. Foremost compensations of nanodiamonds are small and uniform size, nearly spherical shape, large and accessible surface area, and superior mechanical, electric, optical, and thermal properties. In epoxy matrix, these nanoparticles maximize interactions i.e. the interphase formation due to tailorable surface chemistry. Spherical shape of ND maximizes the interphase volume per nanoparticle volume, and so contributes to improved mechanical properties of the nanocomposite. At low concentration, small nanodiamond particle size minimizes the interparticle distance, and develops close contact between the nanoparticles. This results in higher thermal conductivity and better mechanical strength of the resulting epoxy nanocomposite. Type of ND, surface modification, concentration, dispersion, and fabrication technique effectively influences the epoxy/ND composites.

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